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(A) Z=3, N=4; (B) Z=12, N=12; (c) Z=54, 10g = 10/12 mole and contains (10/12)x $0.6022 \times 10^{24} = 0.502 \times 10^{24}$ atomer. When H₂ gas is formed from naturally occurring hydrogen, what percentages of the molecules have molecular weights of approximately 2, 3, and 4? No. restriction in question are W.N., WH or, and "H.H. Let 40) and H.N. Id. M. come of restriction of restriction of restriction what Had and "H. Cormboning above, fuelous Made A. Cormboning above, fuelous consisted, form a charge relience of section. I fuel feachers Mad Law weight 2, W.N. 40). (0.99987" a. 0.9999 89.99%.

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